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1.1.1 Concept of electrochemistry:

Electrical Conduction: The substances are divided into 4 types depending upon their capability of flow of electrons. i)

Conductors: The Substances which allows electricity to pass through them are called conductors. Ex :- Metals, metal sulphides, acids, alkalis, salt sol. and fused salts

ENGINEERING CHEMISTRY

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Electrochemical Potential of a Cell 1) Write two $\frac{1}{2}$ reactions written as reductions 2) Compare E's--the reaction with the highest E is reduced, the other is oxidized 3) Flip the oxidized reaction, or change the sign of E 4) Balance/add the reaction 5) $E_{rxn} = E_{reduced} + E_{oxidized}$

Electrochemistry Notes

Electrochemistry is that branch of chemistry which deals with the study of production of electricity from energy released during spontaneous chemical reactions and the use of electrical energy to bring about non-spontaneous chemical transformations. Importance of Electrochemistry 1. Production of metals like Na, Mg.

Chemistry Notes for class 12

Chapter 3 Electrochemistry

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Electrochemistry is the study of

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Electrochemistry is the study of reactions in which charged particles (ions or electrons) cross the interface between two phases of matter, typically a metallic phase (the electrode) and a conductive solution, or electrolyte. A process of this kind is known generally

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as an electrode process.

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